# Best Available Copy

#### (12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization International Bureau



# 

(43) International Publication Date 28 July 2005 (28.07.2005)

PCT

# (10) International Publication Number WO 2005/068435 A1

(51) International Patent Classification<sup>7</sup>:

(21) International Application Number:
PCT/CZ2004/000088

C07D 239/42

(22) International Filing Date: 17 December 2004 (17.12.2004)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data: PV 2004-86

16 January 2004 (16.01.2004)

(71) Applicant (for all designated States except US): ZEN-TIVA, a. s. [CZ/CZ]; U kabelovny 130, Dolni Mecholupy, 102 37 Praha 10 (CZ).

(72) Inventors; and

- (75) Inventors/Applicants (for US only): SEBEK, Pavel [CZ/CZ]; Radcina 2/521, 161 00 Praha 6 (CZ). RADL, Stanislav [CZ/CZ]; Pertoldova 3380, 143 00 Praha 2 (CZ). STACH, Jan [CZ/CZ]; Slitrova 2006, 190 00 Praha 9 Ujezd nad Lesy (CZ).
- (74) Agents: JIROTKOVA, Ivana et al.; Rott, Ruzicka & Guttmann, Patent & Trademark & Law Office, Nad Stolou 12, 170 00 Praha 7 (CZ).
- (81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD,

MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

#### Declaration under Rule 4.17:

as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PII. PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW, ARIPO patent (BW, GH, GM, KE, LS, MW, MZ, NA. SD. SL. SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW. ML, MR, NE, SN, TD, TG)

#### Published:

with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

DIHVDROXY-6-HEPTENOIC ACID

(57) Abstract: A method of preparation of the hemi-calcium salt of rosuvastatin of formula (I) consists in extracting an aqueous solution of the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6isopropyl-2- [methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid, with optional admixture of sodium or potassium hydroxide or other sodium or potassium salts having inorganic anions, with an organic solvent, incompletely miscible with water, selected from the series of R¹COOR², R¹COR² and R¹OII, wherein R¹ and R² independently represent hydrogen or a residue of a C₁-C₁0 aliphatic hydrocarbon, C6 aromatic hydrocarbon, C5 or C6cyclic hydrocarbon, or a combination of an aliphatic and aromatic or cyclic hydrocarbon, the extract being subsequently shaken with an aqueous solution of an inorganic or C₁-C5 organic calcium salt, and the product of formula I is further isolated by cooling and/or adding an anti-solvent and filtration, and optionally, is converted into its amorphous form.



A method of preparation of hemi-calcium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid

# Technical Field

5

10:

20

25

The invention concerns a new method of preparation of the hemi-calcium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,58)-3,5-dihydroxy-6-heptenoic acid known under the INN name rosuvastatin, formula I.

The mentioned medicament is a prominent representative of hypolipidemic and hypocholesteric pharmaceuticals.

#### 15 Background Art

Rosuvastatin is produced according to the published patent (EP 521471) usually from the sodium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]-pyrimidin-5-yl] (3R,5S)-3,5-dihydroxy-6-heptenoic acid and an appropriate water-soluble calcium salt, preferably from calcium chloride.

The starting sodium salt can be obtained according to the above-mentioned patent from the methyl ester of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]-pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid of formula II via hydrolysis with ethanolic sodium hydroxide or lately (according to international patent application WO 00/49014) from tert-butyl (E)-(6-[2-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl-

(methylsulfonyl)amino]pyrimidin-5-yl]vinyl](4R,6S)-2,2-dimethyl-[1,3]dioxan-4-yl)-acetate of formula III

5 This

10

This intermediate product is first transferred to the corresponding sodium salt by consecutive stirring first with hydrochloric acid and then with sodium hydroxide. The calcium salt is subsequently obtained via addition of calcium chloride to the solution of the sodium salt in water. However, the salt prepared in this way is contaminated with inorganic substances. For example, residual sodium hydroxide reacts with calcium chloride to produce water-insoluble calcium hydroxide. Authors of the new patent application (WO 00/042024) assert that the substance prepared according to patent EP 521471 had an amorphous structure; nevertheless the process of its preparation was difficult to reproduce.

15 According to another patent application (WO 03/016317), the calcium salt can be obtained also via reaction of calcium hydroxide with lactone of formula IV

or other esters of rosuvastatin.

The objective of this invention is to describe a new, improved method of preparation of the hemi-calcium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)-amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid (rosuvastatin), which would not have the mentioned disadvantages, and also an improved method of preparation of the amorphous form.

# Disclosure of Invention

10

15

20

25

The subject matter of the invention consists in an improved method of preparation of the hemiof (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)salt calcium amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid of formula I, wherein an aqueous solution of the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid, with optional admixture of sodium or potassium hydroxide or other sodium or potassium salts having inorganic anions, is extracted with an organic solvent, incompletely miscible with water, selected from the series of R1COOR2, R1COR2 and R1OH, wherein R1 and R2 independently represent hydrogen or a residue of a C<sub>1</sub>-C<sub>10</sub> aliphatic hydrocarbon, C<sub>6</sub> aromatic hydrocarbon, C5 or C6 cyclic hydrocarbon, or a combination of an aliphatic and aromatic or cyclic hydrocarbon, the extract being subsequently shaken with an aqueous solution of an inorganic or C1-C5 organic calcium salt, and the product of formula I is further isolated by cooling and/or adding an anti-solvent and filtration.

The aqueous solution of the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic

5

acid is preferably obtained stepwise by acidic hydrolysis and subsequent alkaline hydrolysis of the protected ester of formula III

or by alkaline opening of the lactone of formula IV

$$\begin{array}{c|c}
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\
 & & \\
 & & \\
 & & & \\
 & & \\
 & & & \\
 & & & \\
 & & & \\
 & & & \\$$

- Extraction of the sodium or potassium salt from the aqueous solution is performed with an ester of formula R<sup>1</sup>COOR<sup>2</sup>, wherein R<sup>1</sup> and R<sup>2</sup> have the above mentioned meanings, or, even more preferably, extraction is made with ester R<sup>1</sup>'COOR<sup>2</sup>', wherein R<sup>1</sup>' and R<sup>2</sup>' are independently hydrogen or a C<sub>1</sub>-C<sub>5</sub> aliphatic residue, preferably with ethyl acetate.
- This whole procedure is based on the surprising finding that the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid can be quantitatively extracted from the aqueous phase into solvents of the type of esters, ketones or alcohols of formulae R<sup>1</sup>COOR<sup>2</sup>, R<sup>1</sup>COR<sup>2</sup> or R<sup>1</sup>OH, wherein R<sup>1</sup> and R<sup>2</sup> have the above-mentioned meaning. The sodium or potassium salt obtained in this way can be quantitatively transferred into the calcium salt by stirring with an aqueous solution of an inorganic or organic calcium salt. Rosuvastatin can be subsequently obtained by evaporation and crystallization.

Another aspect of the invention consists in a new method of preparation of the amorphous form, which is based on dissolving the calcium salt of rosuvastatin in a suitable solvent and adding the same to an anti-solvent, in which rosuvastatin is completely insoluble or little soluble. A solution of the hemi-calcium salt of rosuvastatin in an organic solvent selected from the series of R<sup>1</sup>COOR<sup>2</sup>, R<sup>1</sup>COR<sup>2</sup> or R<sup>1</sup>OH, wherein R<sup>1</sup> and R<sup>2</sup> have the above-mentioned meaning, is added dropwise to an anti-solvent in which rosuvastatin is insoluble, selected from the series including compounds of formulae R<sup>1</sup>H and R<sup>1</sup>OR<sup>2</sup>, wherein R<sup>1</sup> and R<sup>2</sup> have the above-mentioned meaning, or water.

The compound of formula I is dissolved in a solvent preferably selected from the series of R<sup>1</sup>'COOR<sup>2</sup>', R<sup>1</sup>'COR<sup>2</sup>'or R<sup>1</sup>'OH, wherein R<sup>1</sup>' and R<sup>2</sup>' have the above-mentioned meanings, added dropwise to an anti-solvent in which rosuvastatin is insoluble, selected from the series including compounds of formulae R<sup>1</sup>'H, R<sup>1</sup>'OR<sup>2</sup>', wherein R<sup>1</sup>' and R<sup>2</sup>' have the above-mentioned meanings, or water.

15

20

The compound of formula I is preferably dissolved in a solution including ketones, particularly acetone, ethyl methyl ketone, isopropyl methyl ketone, alcohols, particularly methanol, ethanol, isopropanol, or butanols, and further esters, particularly of formic acid, acetic acid or propionic acid with methyl, ethyl or propyl alcohol, and the product is precipitated with solvents including heptane, pentane, cyclohexane, toluene, petroleum ether, diethyl ether or water.

# **Brief Description of Drawings**

25 Figure 1 shows the diffraction pattern of an amorphous sample of the hemi-calcium salt of rosuvastatin.

### Detailed description of the invention

30 Esters of rosuvastatin or rosuvastatin lactone of formula IV can be hydrolyzed in aqueous tetrahydrofuran with sodium hydroxide and the resulting sodium salt of rosuvastatin can be quantitatively extracted into the organic phase, preferably with ethyl acetate. The sodium salt obtained in this way is converted into the calcium salt by shaking a solution of the sodium salt

in ethyl acetate or another solvent of the above-mentioned type with a water soluble calcium salt, preferably calcium acetate. The residual inorganic contaminants are subsequently removed by washing with demineralized water. Evaporation and crystallization can produce rosuvastatin, which is not contaminated with inorganic substances.

5

10

15

20

25

According to the original patent EP 521471, the prepared rosuvastatin had an amorphous structure, but the process is not reproducible. The amorphous form has usually different dissolution characteristics and bio-availability than crystalline forms (Konno T.: Chem. Pharm. Bull. 1990, 38, 2003). In case of rosuvastatin, which is little soluble in water, it is important to have a reproducible process for obtaining the amorphous form.

In our method, it has turned out that perfectly amorphous rosuvastatin can be obtained by dissolving crystalline or semi-crystalline rosuvastatin in a solvent in which rosuvastatin is soluble under cold conditions or at increased temperatures, selected from the series of R<sup>1</sup>COOR<sup>2</sup>, R<sup>1</sup>COR<sup>2</sup> or R<sup>1</sup>OH, wherein R<sup>1</sup> and R<sup>2</sup> have the above-mentioned meaning, and by adding the resulting solution to an anti-solvent in which rosuvastatin is insoluble, selected from the series of R<sup>1</sup>H, R<sup>1</sup>OR<sup>2</sup>, wherein R<sup>1</sup> and R<sup>2</sup> have the above-mentioned meaning, or water. The solvents in which rosuvastatin is soluble under cold conditions or at increased temperatures include those solvents in which solubility is higher than 1 g in 50 ml. Mixtures of suitable solvents can be also used. Examples of such preferable solvents include methanol, ethyl methyl ketone or ethyl acetate. The anti-solvents in which rosuvastatin is insoluble include those in which 1g of the substance does not dissolve in 1,000 ml of the solvent under cold conditions. Examples of such solvents include preferably hexane, pentane, diethyl ether or water. A more detailed list of these solvents has been presented above. The diffraction pattern of a perfectly amorphous sample (prepared according to Example 5) is shown in Fig. 1; the measurements were performed on diffractometer SEIFERT 3000 XRD with a graphite monochromator, radiation CoK $\alpha$  ( $\lambda = 1.790$ Å) within the range  $2.5 - 40^{\circ}2\theta$  with a step 0.03.

The invention is elucidated in more detail in the following examples. The examples, which illustrate preferred alternatives of production of rosuvastatin according to the invention, have a purely illustrative character and do not limit the extent of the invention in any respect. Semi-crystalline rosuvastatin used in Example 5 was obtained according to the original patent EP

521471. Crystalline rosuvastatin used in Examples 6 and 7 was obtained according to WO 00/042024.

#### Examples

5

10

15

## Example 1

Tetrahydrofuran (75 ml) is added to lactone IV (5 g, 10.8 mmol). A solution of 40% NaOH (10 ml) is added during 5 minutes to the solution obtained in this way and the formed heterogeneous mixture is vigorously stirred for 17 h and then poured into a separating funnel containing demineralized water (150 ml) and hexane (50 ml). After shaking, the organic layer is separated and the aqueous layer is extracted with a mixture of hexane (40 ml) and tetrahydrofuran (10 ml). After complete separation, the aqueous layer is extracted with ethyl acetate (1 x 40 ml, 3 x 20 ml). The ethyl acetate extract is then gradually shaken 3 times with demineralized water (5 ml), each containing 1 g of calcium acetate in 5 ml of water. The resulting ethyl acetate extract is washed with demineralized water (2 x 5 ml) and, after drying, is concentrated in a vacuum evaporator to a volume of 30 ml and added dropwise to hexane (150 ml) to give, after filtration, 4.5 g of amorphous rosuvastatin.

<sup>1</sup>H NMR (DMSO) δ:

20 1.22 (d, J = 7, 6H); 1.41 (m, 1H); 1.61 (m, 1H); 2.18 (dd, J = 3, 2H); 3.43 (m, 1H); 3.45 (s, 3H); 3.57 (s, 3H); 3.83 (m,1H); 4.25(m, 1H); 5.56 (dd, J = 7.16, 1H); 6.58 (d, J = 16, 1H); 7.33 (m, 2H); 7.76 (m, 2H)

MS for  $C_{22}H_{28}FN_3O_6SNa [M + Na]^+$ : calculated 504.1; found 503.8.

# 25 Example 2

30

Following the procedure described in Example 1 using potassium hydroxide instead of sodium hydroxide for the hydrolysis of the ester, the corresponding potassium salt of rosuvastatin is obtained. The solution is further treated according to the procedure described in Example 1, to provide 4.2 g of amorphous rosuvastatin.

# Example 3

Tetrahydrofuran (15 ml) is added to ester III (1 g, 1.7 mmol) and after a clear solution is formed, 10% HCl (4 ml) is added. The mixture is stirred for additional 24 hours at ambient temperature. Then, a solution of 40 % NaOH (2 ml) is added to the solution during 5 min and the formed heterogeneous mixture is vigorously stirred for 17 h and then poured into a separating funnel containing demineralized water (30 ml) and hexane (10 ml). After shaking, the organic layer is separated and the aqueous layer is extracted with a mixture of hexane (8 ml) and tetrahydrofuran (2 ml). After complete separation, the aqueous layer is extracted with ethyl acetate (1 x 20 ml, 3 x 10 ml). Combined ethyl acetate extracts are gradually shaken 3 times with demineralized water (1 ml), each containing 0.2 g of calcium acetate in 1 ml of water. The resulting ethyl acetate solution is washed with demineralized water (2 x 3 ml) and after drying with calcium sulfate, it is evaporated in a vacuum evaporator. After crystallization from acetonitrile and water, 0.7 g of rosuvastatin is obtained.

15

20

25

10

5

## Example 4

Tetrahydrofuran (15 ml) is added to ester II (1 g, 2 mmol) and after complete dissolution, a solution of 40 % NaOH (2 ml) is added to the solution over 5 min and the formed heterogeneous mixture is vigorously stirred for 17 h and then poured in a separating funnel containing demineralized water (30 ml) and hexane (10 ml). After shaking, the organic layer is separated and the aqueous layer is extracted with a mixture of hexane (8 ml) and tetrahydrofuran (2 ml). After complete separation, the aqueous layer is extracted with ethyl acetate (1 x 20 ml, 3 x 10 ml). The ethyl acetate solution is subsequently shaken 3 times with demineralized water (1 ml), each containing 0.2 g of calcium acetate in 1 ml of water. The resulting ethyl acetate solution is washed with demineralized water (2 x 3 ml) and evaporated in a vacuum evaporator. After crystallization from acetonitrile and water, 0.7 g of rosuvastatin is obtained.

## 30 Example 5

Semi-crystalline rosuvastatin (1 g) is dissolved in ethyl methyl ketone (10 ml) at 40 °C. After being filtered, the resulting solution is added dropwise to pentane (70 ml), while the mixture is

WO 2005/068435 PCT/CZ2004/000088

vigorously stirred. After 30 min of stirring, the solution is sucked off and dried in vacuo to give 0.95 g of amorphous rosuvastatin.

# Example 6

5

Crystalline rosuvastatin (1.5 g) is dissolved in methanol (10 ml) at 25 °C. After being filtered, the resulting solution is added dropwise to water (150 ml), while the mixture is vigorously stirred at 5 °C. After 30 min of stirring, the solution is sucked off and dried in vacuo to give 1.3 g of amorphous rosuvastatin.

10

# Example 7

Crystalline rosuvastatin (1 g) is dissolved in methanol (10 ml) at 25 °C. After being filtered, the resulting solution is added dropwise to diethyl ether (150 ml) at 25 °C. After 30 min of stirring, the solution is sucked off and dried in vacuo to give 0.7 g of amorphous rosuvastatin.

# CLAIMS

1. A method of preparation of the hemi-calcium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid of formula I, i.e. rosuvastatin

characterized in that an aqueous solution of the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid, with optional admixture of sodium or potassium hydroxide or other sodium or potassium salts having inorganic anions, is extracted with an organic solvent, incompletely miscible with water, selected from the series of R<sup>1</sup>COOR<sup>2</sup>, R<sup>1</sup>COR<sup>2</sup> and R<sup>1</sup>OH, wherein R<sup>1</sup> and R<sup>2</sup> independently represent hydrogen or a residue of a C<sub>1</sub>-C<sub>10</sub> aliphatic hydrocarbon, C<sub>6</sub> aromatic hydrocarbon, C<sub>5</sub> or C<sub>6</sub> cyclic hydrocarbon, or a combination of an aliphatic and aromatic or cyclic hydrocarbon, the extract being subsequently shaken with an aqueous solution of an inorganic or C<sub>1</sub>-C<sub>5</sub> organic calcium salt, and the product of formula I is further isolated by cooling and/or adding an anti-solvent and filtration, and, optionally, it is converted into its amorphous form.

20

25

10

15

2. The method according to claim 1 characterized in that the aqueous solution of the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl-(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid is obtained stepwise by acidic hydrolysis and subsequent alkaline hydrolysis of the protected ester of formula III

10

15

20

or by alkaline opening of the lactone of formula IV

- 3. The method according to claim 1 characterized in that the extraction of the sodium or potassium salt from the aqueous solution is performed with an ester of formula  $R^1COOR^2$ , wherein  $R^1$  and  $R^2$  are as defined in claim 1.
- 4. The method according to claim 1 characterized in that the extraction is performed with ester  $R^{1'}COOR^{2'}$ , wherein  $R^{1'}$  and  $R^{2'}$  are independently hydrogen or a  $C_1$ - $C_5$  aliphatic residue, preferably with ethyl acetate.

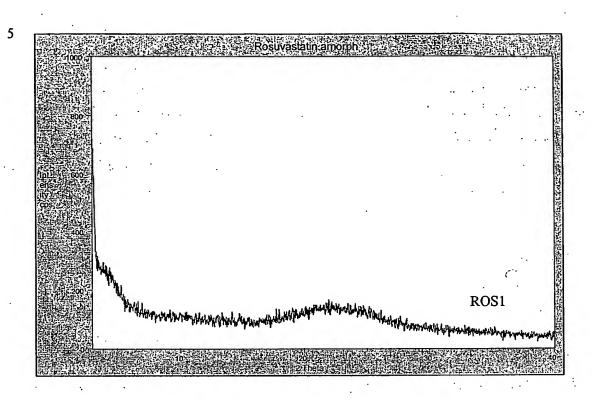
5. A method of the preparation of the amorphous form of the hemi-calcium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid of formula I, i.e. rosuvastatin, according to claim 1, characterized in that a solution of the hemi-calcium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid in an organic solvent selected from the series of  $R^1COOR^2$ ,  $R^1COR^2$  and  $R^1OH$ , wherein  $R^1$  and  $R^2$  are as defined in claim 1, is added dropwise to a solvent in which

rosuvastatin is insoluble, selected from the series including compounds of formulae  $R^1H$  and  $R^1OR^2$ , wherein  $R^1$  and  $R^2$  are as defined in claim 1, and water.

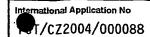
- 6. The method according to claim 5 characterized in that the compound of formula I is dissolved in a solvent selected from the series of  $R^{1'}COOR^{2'}$ ,  $R^{1'}COR^{2'}$  and  $R^{1'}OH$ , wherein  $R^{1'}$  and  $R^{2'}$  are as defined in claim 4, is added dropwise to a solvent in which rosuvastatin is insoluble, selected from the series including compounds of formulae  $R^{1'}H$  or  $R^{1'}OR^{2'}$ , wherein  $R^{1'}$  and  $R^{2'}$  are as defined in claim 4, and water.
- 7. The method according to claim 5 characterized in that the compound of formula I is dissolved in a solvent including ketones, particularly acetone, ethyl methyl ketone, isopropyl methyl ketone, alcohols, particularly methanol, ethanol, isopropanol, or butanols, further esters, particularly of formic acid, acetic acid or propionic acid with methyl, ethyl or propyl alcohol, and the product is precipitated with solvents including heptane, pentane, cyclohexane, toluene, petroleum ether, diethyl ether or water.

1/1

Fig. 1



# INTERNATIONAL SEARCH REPORT



A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C07D239/42							
According to International Patent Classification (IPC) or to both national classification and IPC							
B. FIELDS SEARCHED							
Minimum documentation searched (classification system followed by classification symbols)  IPC 7 C07D							
Documentation searched other than minimum documentation to the extent that such documents are included in the field is searched							
Documentation searched other than minimum documentation to the extent that such documents are included in the least searched							
Electronic data base consulted during the International search (name of data base and, where practical, search terms used)							
EPO-Internal, WPI Data, CHEM ABS Data							
C. DOCUMENTS CONSIDERED TO BE RELEVANT							
Category °	Citation of document, with indication, where appropriate, of the rele	evant passages	Relevant to claim No.				
A	WO 03/016317 A (TEVA PHARMACEUTIC INDUSTRIES LTD; TEVA PHARMACEUTIC INC; NID) 27 February 2003 (2003- cited in the application page 13, line 1 - page 15, line 2	1					
A	WO 00/42024 A (ASTRAZENECA UK LIM TAYLOR, NIGEL, PHILLIP) 20 July 2000 (2000-07-20) cited in the application example 1	IITED;	1				
Α .	EP 0 521 471 A (SHIONOGI SEIYAKU KAISHA) 7 January 1993 (1993-01-0 cited in the application example 7		1				
Furti	Further documents are listed in the continuation of box C.  Y Patent family members are listed in annex.						
© Special categories of cited documents :							
*Special categories of cited documents:  *T* later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention  *E* earlier document but published on or after the international  *Y* document of particular relevance: the claimed invention							
tiling d	late	"X" document of particular relevance; the c cannot be considered novel or cannot involve an inventive step when the do	be considered to				
*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) involve an inventive step when the document is taken alone document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the							
*O* document referring to an oral disclosure, use, exhibition or other means document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.							
	ent published prior to the international filing date but nan the priority date claimed	& document member of the same patent family					
Date of the actual completion of the international search		Date of mailing of the international sea	rch report				
2	0 April 2005	27/04/2005					
Name and mailing address of the ISA  European Patent Office, P.B. 5818 Patentlaan 2		Authorized officer					
NL – 2280 HV Rijswijk Tel. (+31–70) 340–2040, Tx. 31 651 epo nl, Fax: (+31–70) 340–3016		Fanni, S					

# INTERNATIONAL SEARCH REPORT

T/CZ2004/00088

					1-01/0/	Fe1/CZ2004/000088	
	Patent document cited in search report		Publication date		Patent family member(s)	Publication date	
	WO 03016317	Α	27-02-2003	US	2002099224 A1	25-07-2002	
ı				CA	2450820 A1	27-02-2003	
				CN	1543468 A	03-11-2004	
				CZ	20040337 A3	12-01-2005	
				EP	1425287 A1	09-06-2004	
				HR	20040255 A2	31-08-2004	
ŀ				JP	2005500382 T	06-01-2005	
				NZ	529913 A	24-03-2005	
				SK	1402004 A3	03-01-2005	
				TR	200302281 T2	21-09-2004	
			•	WO	03016317 A1	27-02-2003	
				US	2003114685 A1	19-06-2003	
				US	2004176615 A1	09-09-2004	
	WO 0042024	Α	20-07-2000	AT	282027 T	15-11-2004	
				ΑU	762909 B2	10-07-2003	
1				ΑU	1882600 A	01-08-2000	
				BR	9916786 A	16-10-2001	
				CA	2356212 A1	20-07-2000	
				CN	1333756 A	30-01-2002	
				CZ	20012460 A3	17-10-2001	
				DE	69921855 D1	16-12-2004	
				EE	200100359 A	16-12-2002	
				EP	1144389 A1	17-10-2001	
		·		WO	0042024 A1	20-07-2000	
				HÚ	0104828 A2	29-07-2002	
				ID JP	29432 A 2002539078 T	30-08-2001	
1				NO	2002539078 T 20013368 A	19-11-2002 05-09-2001	
				NZ	512560 A	29-08-2003	
1				PL	348775 A1	17-06-2002	
				RŪ	2236404 C2	20-09-2004	
				SK	9632001 A3	03-12-2001	
1				TR	200101894 T2	21-12-2001	
				ÜS	2004009997 A1	15-01-2004	
				ÜS	6589959 B1	08-07-2003	
				ZA	200105187 A	23-09-2002	
	EP 0521471	A	07-01 <b>-</b> 1993	AT	197149 T	15-11-2000	
	_:	••	2. 01 1000	CA	2072945 A1	02-01-1993	
				CY	2226 A	18-04-2003	
				DE	69231530 D1	30-11-2000	
				DE	69231530 T2	13-06-2001	
				DK	521471 T3	05-02-2001	
				EP	0521471 A1	07-01-1993	
				ES	2153824 T3	16-03-2001	
				GR	3035189 T3	30-04-2001	
				HK	1011986 A1	13-07-2001	
				HU	220624 B1	28-03-2002	
				HU	61531 A2	28-01-1993	
				JP	2648897 B2	03-09-1997	
			•	JP	5178841 A	20-07-1993	
				KR	9605951 B1	06-05-1996	
				LU	91042 A9	24-11-2003	
				NL	300125 I1	01-07-2003	
				PT	521471 T	30-04-2001	
l				US	RE37314 E1	07-08-2001	
				US	5260440 A	09-11-1993	
	A/210 (patent family annex) (Janua						

# This Page is Inserted by IFW Indexing and Scanning Operations and is not part of the Official Record

# **BEST AVAILABLE IMAGES**

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images include but are not limited to the items checked:			
☐ BLACK BORDERS			
☐ IMAGE CUT OFF AT TOP, BOTTOM OR SIDES			
☐ FADED TEXT OR DRAWING			
☐ BLURRED OR ILLEGIBLE TEXT OR DRAWING			
☐ SKEWED/SLANTED IMAGES			
☐ COLOR OR BLACK AND WHITE PHOTOGRAPHS			
☐ GRAY SCALE DOCUMENTS			
LINES OR MARKS ON ORIGINAL DOCUMENT			
☐ REFERENCE(S) OR EXHIBIT(S) SUBMITTED ARE POOR QUALITY			
Потитр.			

# IMAGES ARE BEST AVAILABLE COPY.

As rescanning these documents will not correct the image problems checked, please do not report these problems to the IFW Image Problem Mailbox.